

6.3 (1) Extra Practice

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| 1. Iron (II) chlorate | 17. LiNO_3 | 32. Nitrogen pentafluoride |
| 2. Sodium carbonate | 18. $\text{Cu}(\text{OH})_2$ | 33. KCl |
| 3. Chromium (II) Oxide | 19. Copper (III) hydroxide | 34. SO |
| 4. Dicarbon trioxide | 20. Li_2SO_4 | 35. P_3Br_2 |
| 5. Lithium sulfide | 21. Sodium nitride | 36. Carbon dioxide |
| 6. Carbon monoxide | 22. NH_4OH | 37. Copper (II) chloride |
| 7. C_4F_1 | 23. Beryllium phosphide | 38. Lithium nitride |
| 8. $(\text{NH}_4)_2\text{O}$ | 24. Na_2S | 39. $\text{Fe}(\text{C}_2\text{H}_3\text{O}_2)_3$ |
| 9. Ammonium chloride | 25. Disulfur heptachloride | 40. Potassium hydroxide |
| 10. Cu_2O_3 | 26. Magnesium nitride | 41. NH_4OH |
| 11. Carbon trioxide | 27. $\text{Cu}_2(\text{SO}_4)_3$ | 42. Aluminum sulfide |
| 12. $\text{Ca}_3(\text{PO}_4)$ | 28. Hexacarbon heptabromide | 43. Copper (III) sulfide |
| 13. Copper (III) carbonate | 29. Iron (III) phosphide | 44. CuF_2 |
| 14. FePO_4 | 30. $\text{CaC}_2\text{H}_3\text{O}_2$ | 45. Mg_3N_2 |
| 15. Calcium oxide | 31. Magnesium sulfide | 46. FeSO_4 |